

IPRO 304-A: Compact Climatization Units for Automotive Applications

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Arthur Anderson, Tamara Brooks,
Janet Foster, Suneeta Julius,
Matthew Luebbers, Vickie Mark,
Eli Milam, Muhammad Salameh,
Evelyne Schranz, Cari
Youngblood

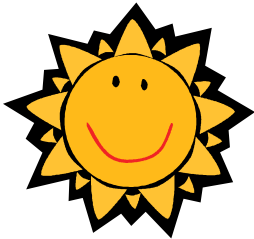


Introduction & Background

- On a hot, sunny day, cars can become unbearably hot.
- The temperature increases due to solar radiation.
- This radiation can be used as an energy source to cool the car.



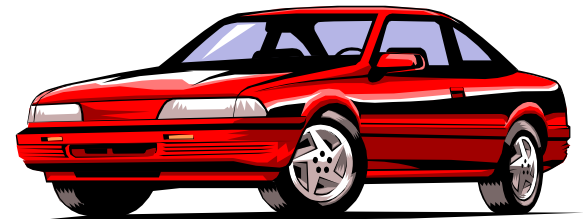
Defining the System



Surroundings:

Temperature: 100 °F

Pressure:
atmospheric



Car:

Temperature: 120 °F

Pressure: atmospheric

$T_{\text{DESIRED}} = 90 \text{ °F}$

Calculation of Heat load

- $P(\text{atm}) = 1.00 \text{ atm}$
- $n = PV/RT = 108 \text{ moles air}$
- $\rho = MWn/V = 1.10 \text{ kg/m}^3$
- $m = MW*n = 3.13 \text{ kg air}$
- $Q_{\text{HL}} = mC_p(T_{\text{CAR}} - T_{\text{DESIRED}}) = 52 \text{ kJ}$

Solar Radiation

- Insolation
 - Variables: S, Zenith angle (Z), Latitude (Φ), Hour angle (H), Solar declination angle (δ)
- Radiation heat load
 - Variables: Angle from horizontal (θ), Emissivity (ϵ)

Φ	δ	Time	H	Z	I (W/m ²)	Description	I _{window (45° from horizontal)}	Radiation (W/m ²)
33.4	23.5	8	-60.0	53.0	602	Phoenix, summer, 8am	426	400
33.4	23.5	12	0.0	9.9	985	Phoenix, summer, noon	697	655
33.4	23.5	17	75.0	65.3	418	Phoenix, summer, 5pm	295	278

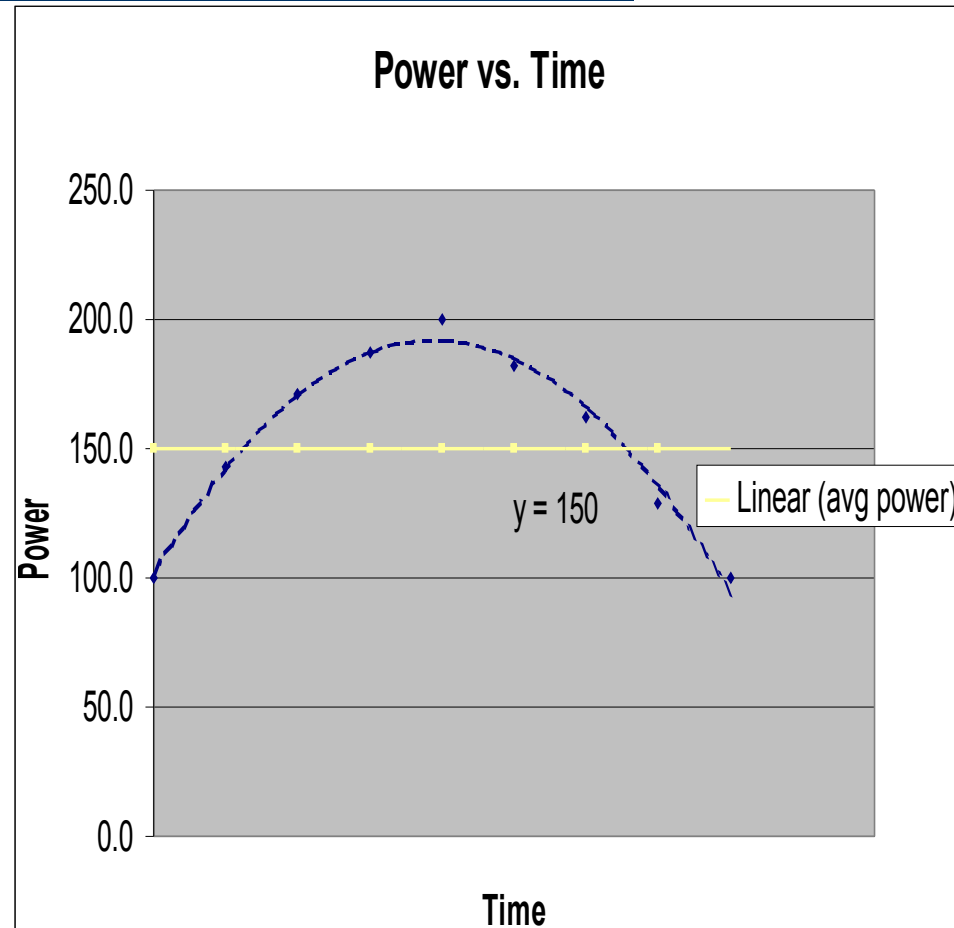
Power System

- Solar Panel
 - Black body radiation
 - Windshield has a non-unit emissivity.
 - Aesthetics
- Battery (to regulate the solar panel)
 - Primary
 - Secondary



Power System: Description

- Solar radiation is variable
- Average power is 150 Watts
- Choices
 - NiCd, NiZn, Lead Acid, Li+
 - Memory Effect
 - Operating Temperature



Power System: Batteries

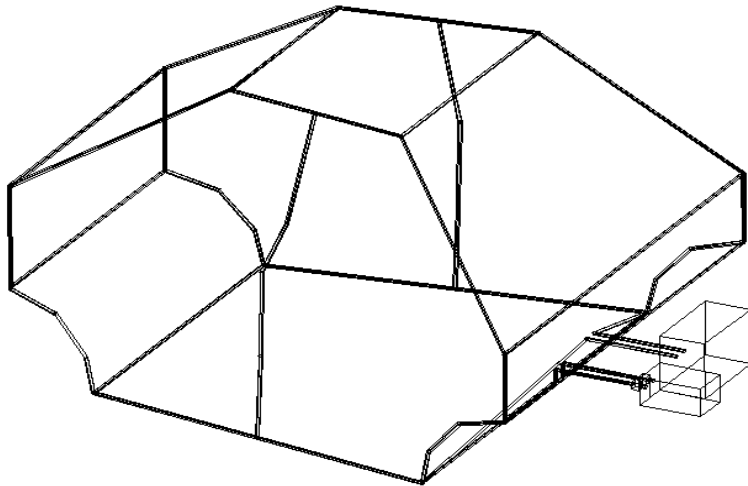
- What are they used for?
 - To power the compressor or freezer
- What is the voltage?
 - Approximately 12V
- For how long can the battery run?
 - 4 hours
- What is the current needed?
 - $I = P/V \rightarrow 150 \text{ Watts} / 12 \text{ Volts} = 12.5 \text{ A}$
 - $12.5 \text{ A} \times 4 \text{ hr} = 50 \text{ Amp-hr}$

Power System: Battery Choice

- NiZn
 - No Memory Effect
 - High Energy Density
 - 60 to 80 W h/kg
 - High Cycle Life
 - 500
 - Reasonable Cost
 - \$30/cell

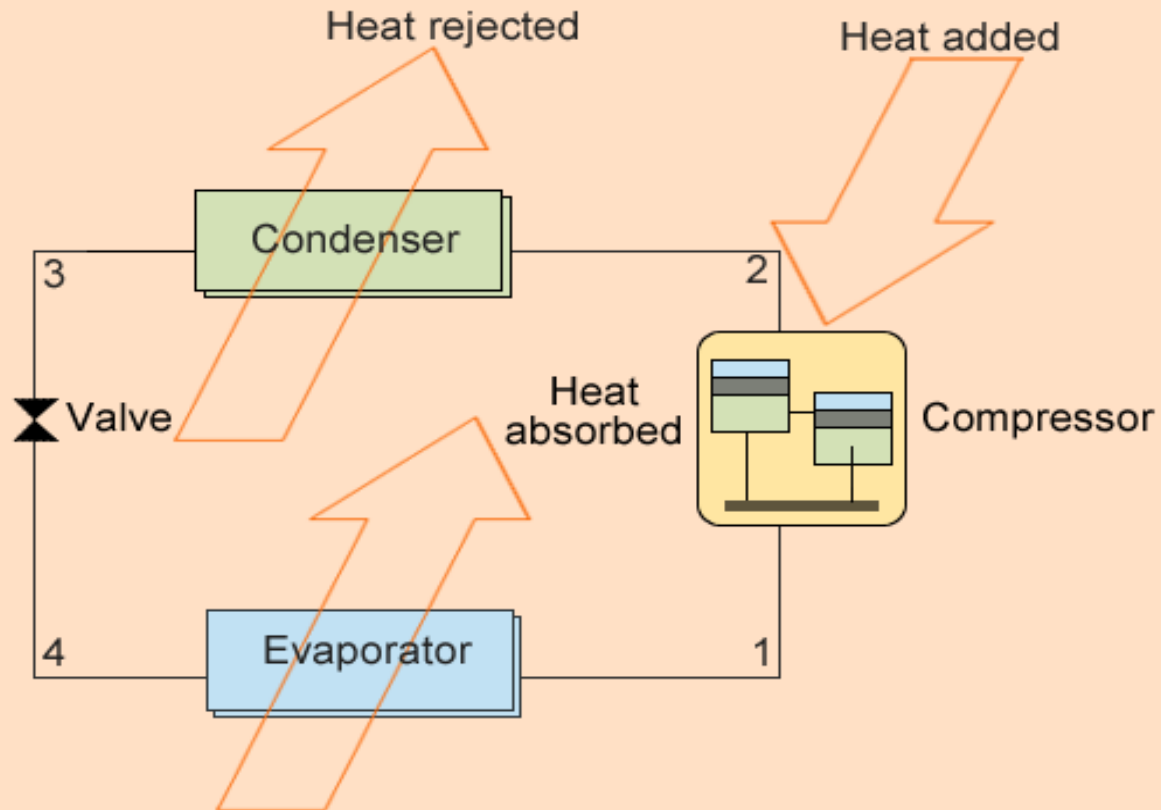


Available Space



- This image represents the basic frame that is available for the refrigeration / heat pipe systems
- Note the positioning of the cooling chest and the compressor in the trunk of the car
- This is an isometric view from the rear driver's side of the car

Strategy 1: Refrigeration Cycle



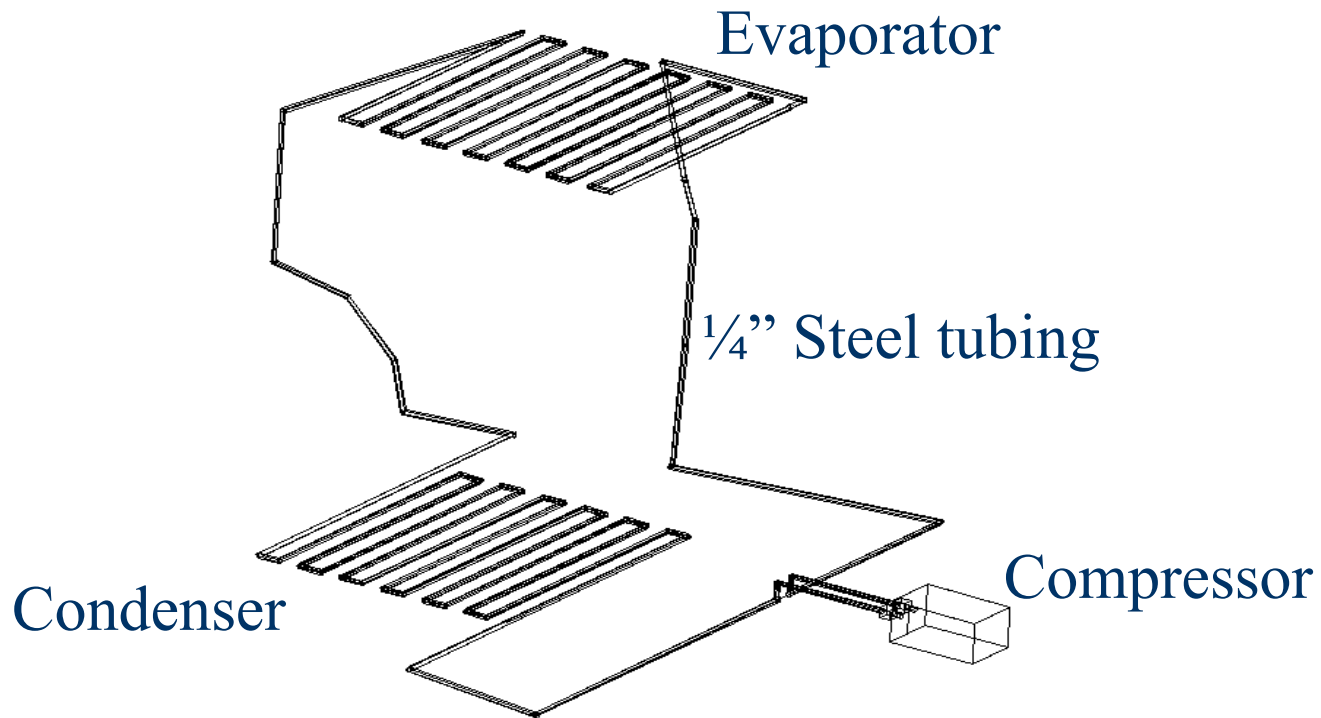
Refrigeration: Description

- Refrigerant: Ethyl chloride
- Pressure in and out of evaporator: 1.361 atm
- Vapor pressure of Ethyl chloride @68.31°F is 1.36 atm

Refrigeration: Description

- Heat load absorbed into evaporator in 15 min.:
735.1 Watts (44.11 KJ)
- Compressor work: 150.0 Watts
- Pressure out of compressor: 4.590 atm
- Temp into condenser: 187.5 °F
- Temp out of condenser: 140.0 °F

Refrigeration: System Layout



Refrigeration: Specs

- Evaporator:
 - $T_{IN} = 68.313 \text{ F}$
 - $T_{OUT} = 68.314 \text{ F}$
 - Heat Duty = 735.1 W
- Condenser:
 - $T_{IN} = 187.5 \text{ F}$
 - $T_{OUT} = 140 \text{ F}$
 - Heat Duty = 885.1 W

Refrigeration: Heat transfer area

- $\Delta T_{LM} = (T_2 - T_1) / \ln(T_2 / T_1)$
 - $\Delta T_{LM, EVAP} = 293.3242 \text{ K}$
 - $\Delta T_{LM, COND} = 346.1768 \text{ K}$
- $Q = U * A * \Delta T_{LM}$
- Heat Transfer Coefficient = $2.8 \text{ W/m}^2 * \text{K}$
 - (Minimum value for still air)
- $A_{EVAP} = Q_{EVAP} / U * \Delta T_{LM, EVAP} = 0.895 \text{ m}^2$
- $A_{COND} = Q_{COND} / U * \Delta T_{LM, COND} = 0.913 \text{ m}^2$

Length of Steel Tubing

- $V = \pi r^2 L$
- $V_{\text{TOTAL}} = 2.02$ Liters Ethyl Chloride
- $A = 2\pi rL$
- $L_{\text{EVAP}} = 26.86$ m
- $L_{\text{COND}} = 27.95$ m
- $L_{\text{STRAIGHT}} = 8$ m
- Total Length of 0.21 inch Radius
 $\cong 63$ m



Tubing

Refrigeration: Costing

Equipment	Size Range	Capacity Unit	Cost
Compressor	150	Watts	\$119.15
Valve	0.02	meters	\$6.50
Solar Panel	150	Watts	\$700.00
Ni-Zn Battery	12 x 2	Volts	\$400.00
Stainless steel tube	63	meters	\$339.00
Ethyl Chloride	2000	Grams	\$141.50
		Total	\$1706.15

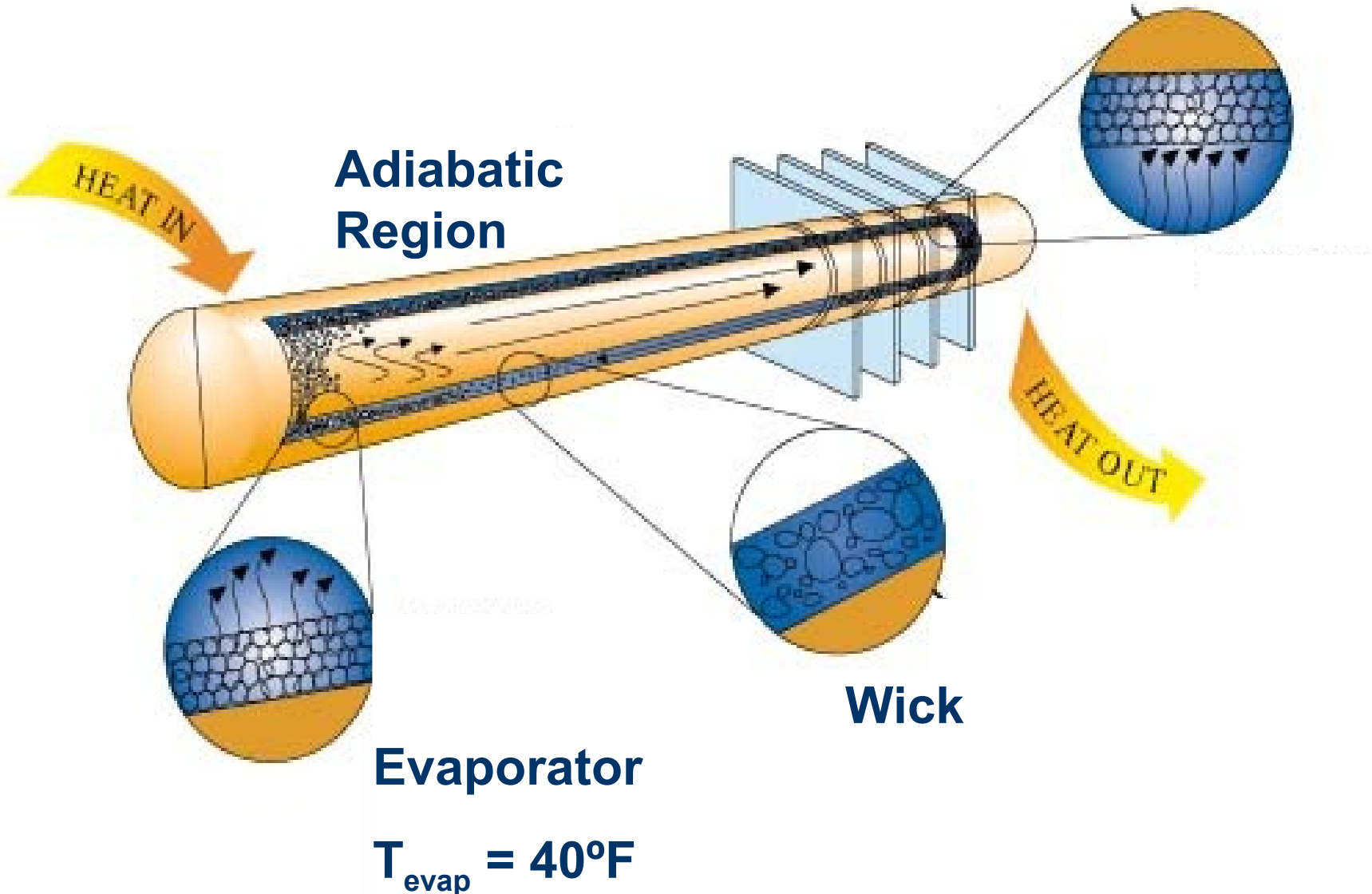
Refrigeration: Safety

- Ethyl Chloride:
 - Toxicity Levels: 30,000 μ g/m³
 - Effects: Simple asphyxiant; anesthetic effect in high concentrations; forms toxic substances upon heating of combustion.
 - Corrosive in moist conditions; need to use stainless steel material.

Heat Pipes

Condenser

$$T_{\text{cond}} = 83^{\circ}\text{F}$$



Heat Pipes: Cooling Source

- Cooling/freezing ability from 40°F to 0°F
- Inside Dimensions:
L 13.75" x W 7.5" x H 7.5"
- Power consumption:
0.9 - 3.9 Amps on a
12-Volt DC adapter



Heat Pipes: Cooling Source

- Power consumption: 3.0 Amps in refrigerator
12Volt AC adapter
- Cooling/freezing ability from 75°F to 25°F
- Self-contained and sealed against the environment
- Inside Dimensions: L 8.7" x W 5.3" x H 2.6"



Ivory

Gray

Red

Dark Blue

Woodgrain

Heat pipes: Working fluid

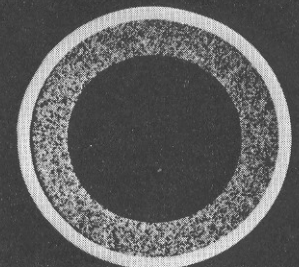
- Key properties
 - High latent heat
 - High surface tension
- Possible fluids

Fluid	Boiling point at 1 atm (°C)	Surface tension, σ , at 50 °C (N/m)	Liquid density, ρ , at 50 °C (kg/m ³)	Liquid viscosity, μ , at 50 °C (cP)	Latent heat, L, at 50 °C (kJ/kg)
R-11 (CCl ₃ F)	24	0.0153	1414	0.35	172
Pentane (n-C ₅)	28	0.0127	596.0	0.187	348.9

Heat pipes: Wick material

- Pore size vs. permeability

Wick material and mesh size	Pore radius, r (mm)	Wire diameter, d_w (mm)	Porosity, ϵ (%)	Permeability, K (m ²)
Stainless steel, 403 mesh	0.019	0.025	36	1.9×10^{-11}
Stainless steel, 101.6 mesh	0.08	0.09	41	4.3×10^{-10}
Copper powder, sintered, 45-56m	0.009		52	1.74×10^{-12}



Heat pipes: Heat flow

- From pressure balance, $(\Delta P_c)_{max} = \Delta P_l + \Delta P_g$

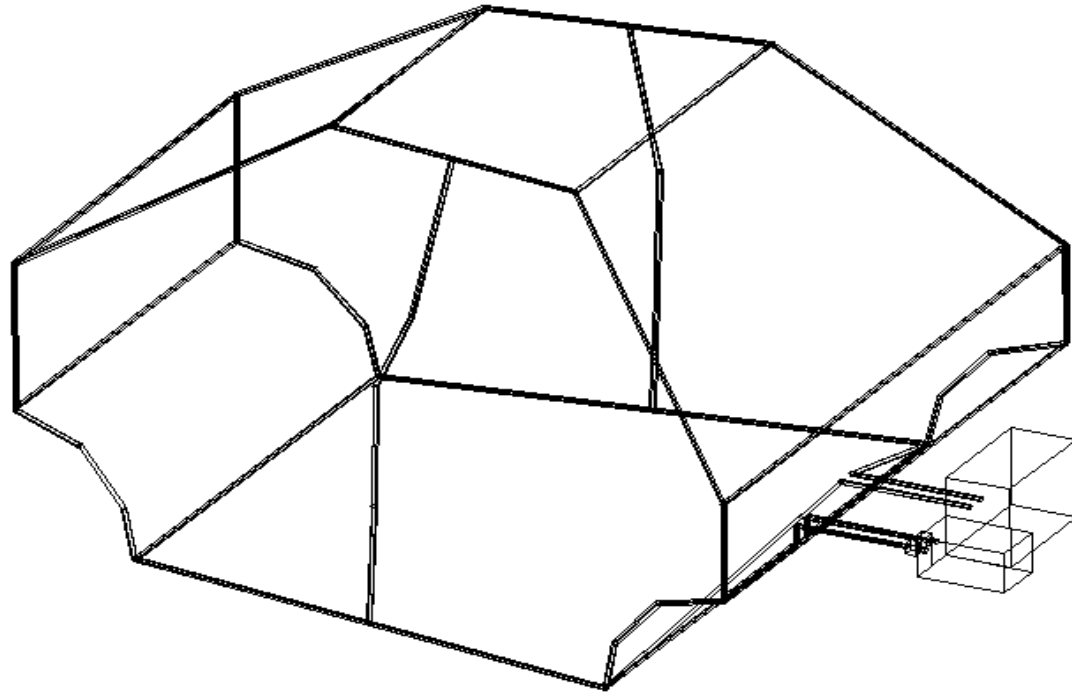
- Mass flow rate,

$$m = \rho_l A_w K / \mu l_{eff} * (2\sigma / r - \rho_l g l_{eff} \sin\phi)$$

- Heat flow = mass flow * Latent heat

Description	m (kg/s)	Q (W)	W/1.5m ²	Total W	Layers
4 layers 400 ss wick, ga, 45°	2.9E-06	1.05	157	471	3
6 layers 400 ss wick, ga, 45°	4.3E-06	1.57	236	707	3
8 layers 400 ss wick, ga, 45°	5.8E-06	2.09	314	628	2
10 layers 400 ss wick, ga, 45°	8.7E-06	3.14	471	943	2

Placement of Heat pipes



Heat Pipes: Costing

Equipment	Size Range	Capacity Unit	Number	Cost
Aluminum pipe	0.4	meters	300	\$1594.49
Stainless steel Mesh	0.915 x 30.49	meters		\$7.00
n-pentane	0.00379	m ³		\$50.00
Cooler	0.22 x 0.134 x 0.066	m	1	\$400.00
			Total	\$2051.49

Heat Pipes: Safety

- n-Pentane
 - Extremely flammable liquid and vapor
 - Vapor may cause flash fire
 - Harmful or fatal if swallowed or inhaled
 - Affects central nervous system
 - Causes irritation to skin, eyes and respiratory tract
 - Toxicity in 8-Hour Time Weighted Average (TWA), 1000 ppm

Results and Conclusions

- In the refrigeration system, 735.1 Watts of heat is being extracted while only 655 Watts are coming into the car.
- It is possible to design the heat pipe system to match the heat load. It would require 300 – 400 heat pipes of ~ 2 Watts each.
- Both systems are feasible but not cost efficient.

Recommendations

- Find a Better Refrigerant.
- Find a Less Expensive Battery.
- Better Integration of the Refrigeration Cycle in the Car's Frame.
- Further optimize the heat pipe system.