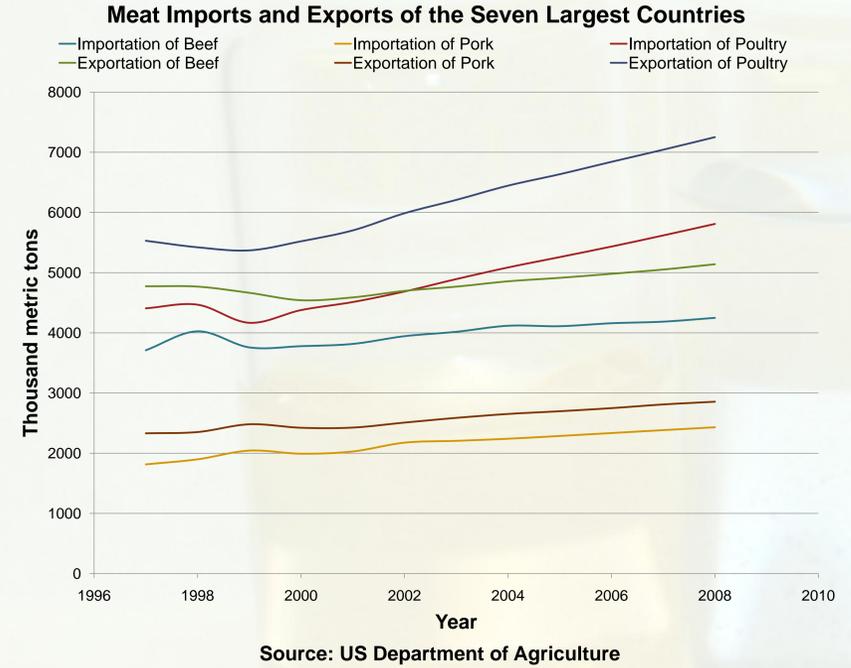


## Problem



• Maintaining the quality and safety of perishable items during shipping and storage is greatly dependent on maintaining adequate thermal conditions. How can this temperature be monitored effectively and efficiently?



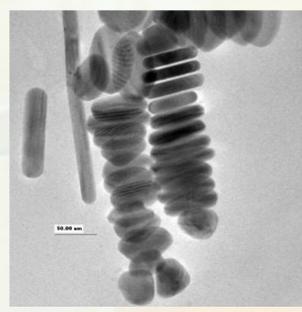
• A cost-effective thermal-history indicator could greatly improve quality assurance and safety of perishable products.

## Opportunity

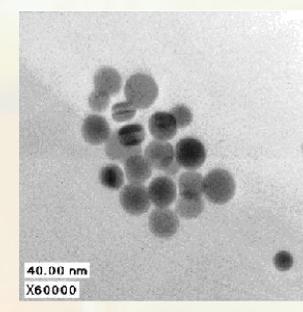
- Because of their shape dependent optical properties and morphological instability, **silver nanorods** could be used as a convenient and cost-efficient thermal indicator to determine if packages have not been stored under adequate thermal conditions.
- If a **continuous process** for producing **silver nanorods** could be established, the **cost** could be sharply **reduced**.

## Background

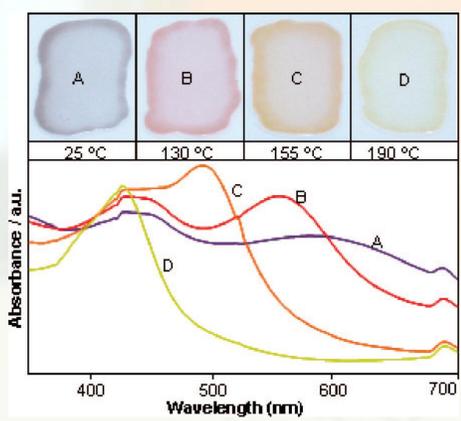
- Nanoparticles are smaller than the size of a dust particle.
- It has a unique **color changing property** which is directly dependant on **changes in both temperature and time**. This color change is due to morphological changes of nanoparticles.
- One significant application involves the use of **silver nanorods** as **thermal history indicators** as they have the unique property of changing color as ambient temperature changes.
- Work is being done on the **development** of a **continuous flow process** for large scale and **commercial use** of silver nanorods.



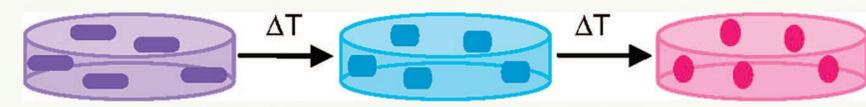
Silver nanorods after one week



Silver nanorods after 4 weeks



Nanorod Color and Absorption Changes

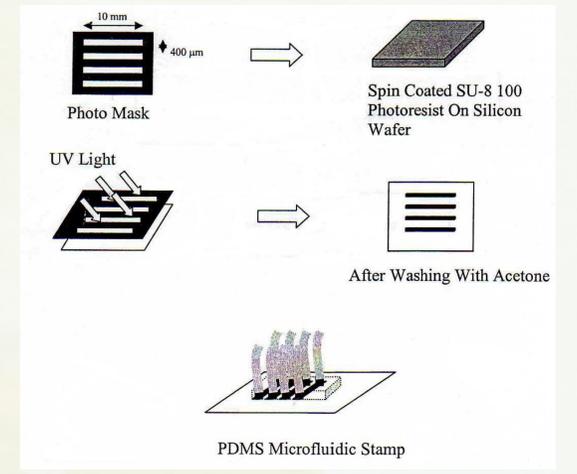


Artist's Conception of Nanorod Shape and Color Changes

Source: Tollan, C.; et al. /ACS Appl. Mater. Interfaces/ 2009, 1, 348

## Objectives

- Optimize the existing batch process for creating silver nanorods.
- Develop a continuous process for producing silver nanorods.
- Explore the use of silver nanorods as thermal indicators and prove the economic viability of the nanorods
- Tailor silver nanorods for their use at different conditions of thermal history.
- Design and build a microreactor for continuous nanorod production.



Microfluidics Channel Design



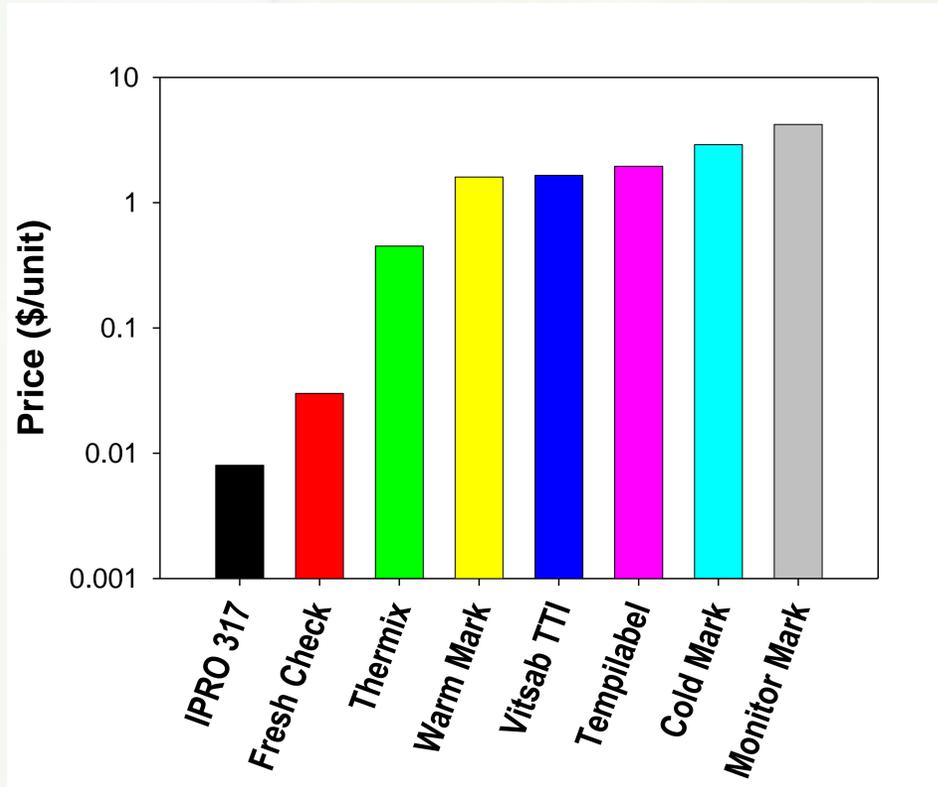


## Methodology

The team was divided into the following groups.

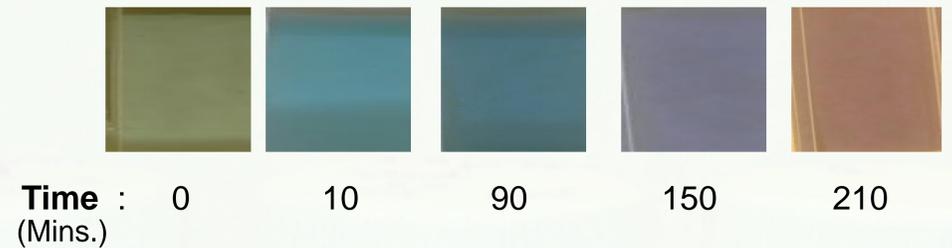
- **Research Team:** Compiled information about the batch process presently being used for the production of silver nanorods and also the feasibility of a continuous process.
- **Lab Team:** Conducted experiments using the batch process to make silver nanorods and designed a method to convert the process from batch to continuous with a microfluidic reactor.
- **Economics Team:** Researched the cost and benefits of converting the process from batch to continuous process and concluded that it was economically feasible (see below).

## Economics

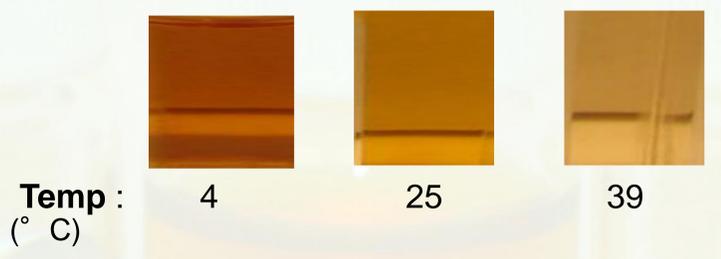


• The Price axis is set at a logarithmic scale to demonstrate the difference in costs by order of magnitudes (factors of 10).

## Results



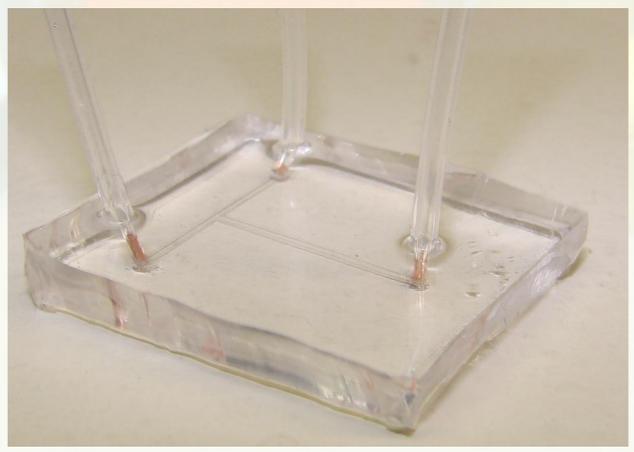
Silver Nanorod color change over Time (held at 60° C)



Nanorod Color Change with Temperature ( after 7 days )



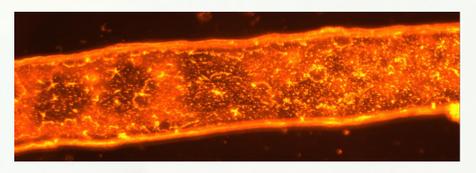
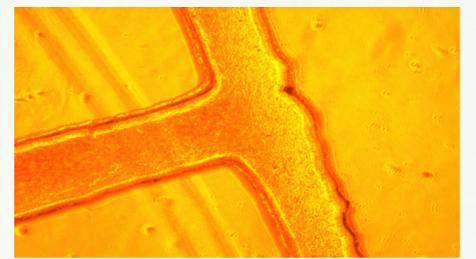
Nanorod Color Change with Temperature ( after 13 days )



Microfluidic Model

- Microfluidic channels were developed as the first step towards a continuous process of producing nanorods.
- It has two inlets on one end for the reactant solutions and an outlet on the other end for the produced nanorods along with solution.

- Microscopic (10X) view of microfluidic channel where two inlets connect (Diameter 200µm)
- Microscopic (20X) view of microfluidic channel where diffusion occurs across concentration gradient (Diameter 200µm)



## Conclusion

- Silver nanorods were successfully produced using a batch process.
- The results clearly demonstrate the wide spectrum of color that nanoparticles use to indicate a wider range of temperature relative to its competitors.
- At 0.8 cents/label, the cost of silver nanorods as thermal indicators was determined to be economically viable.
- A continuous flow process using a microfluidic model was designed and testing is in progress to determine where solution mixing occurs through diffusion.

## Future Works

- Further develop the microfluidic model to incorporate a steady outlet stream of nanoparticles.
- Explore the viability of introducing reactants like iodine to lower the diffusion time in the reactor.
- Silver nanoparticles' antimicrobial uses could be explored to develop pharmaceutical tools to combat drug-resistant viruses and bacteria in human beings.

## Acknowledgments

- Dr. Victor Perez-Luna, Advisor
- Sudipto Chakraborty, Advisor