



Enpro 356

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## » The Problem

Mercury is a **highly toxic** heavy metal that poses a major public **health threat**. Because mercury can **interfere with developmental process**, fetuses and children are most likely to be at risk.

## » Emission Sources

### Anthropogenic Sources

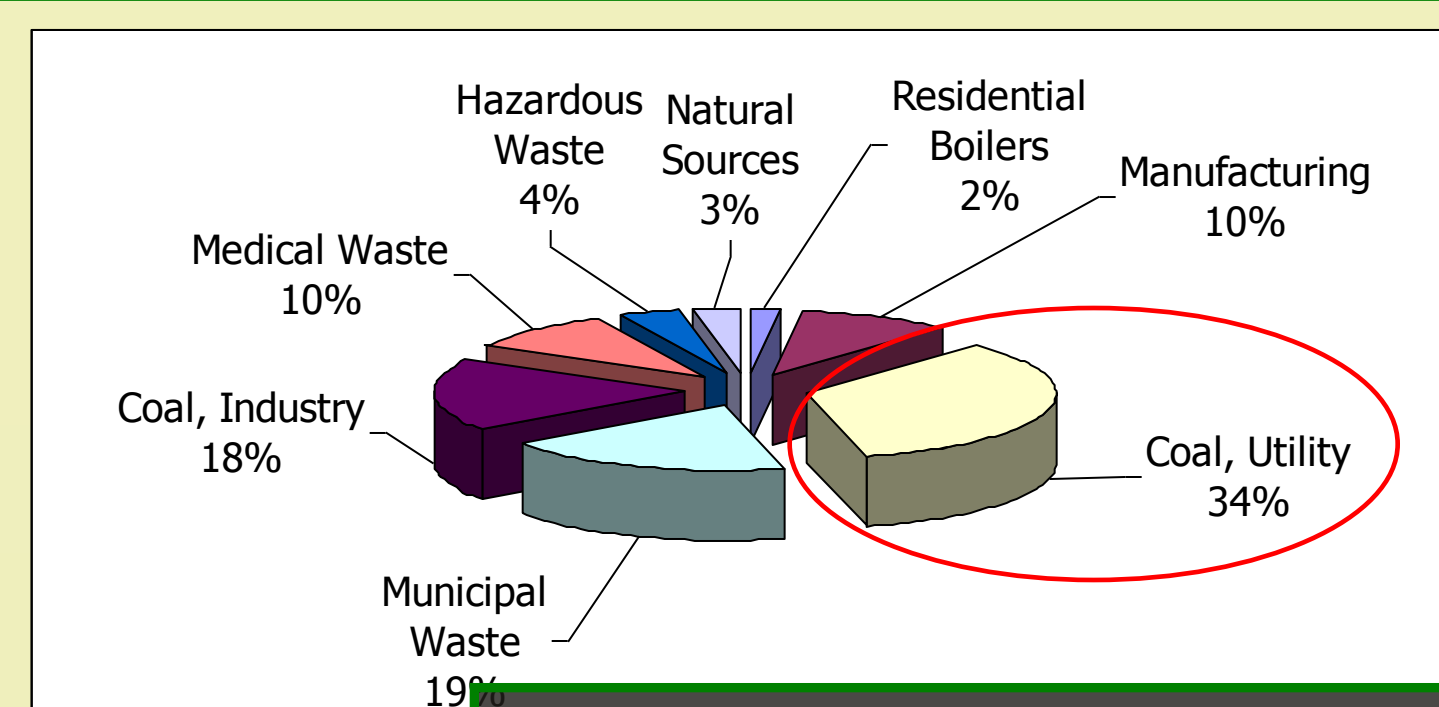
- Coal-fired power plants account for about 34% of mercury emissions in the U.S. -- by far the largest single source.
- U.S. electric utilities release approximately 48 tons of mercury every year.
- Manmade processes such as gold and ore mining
- Medical waste incineration, municipal and hazardous waste combustion
- Cement manufacturing and pulp and paper milling.

### Natural Sources

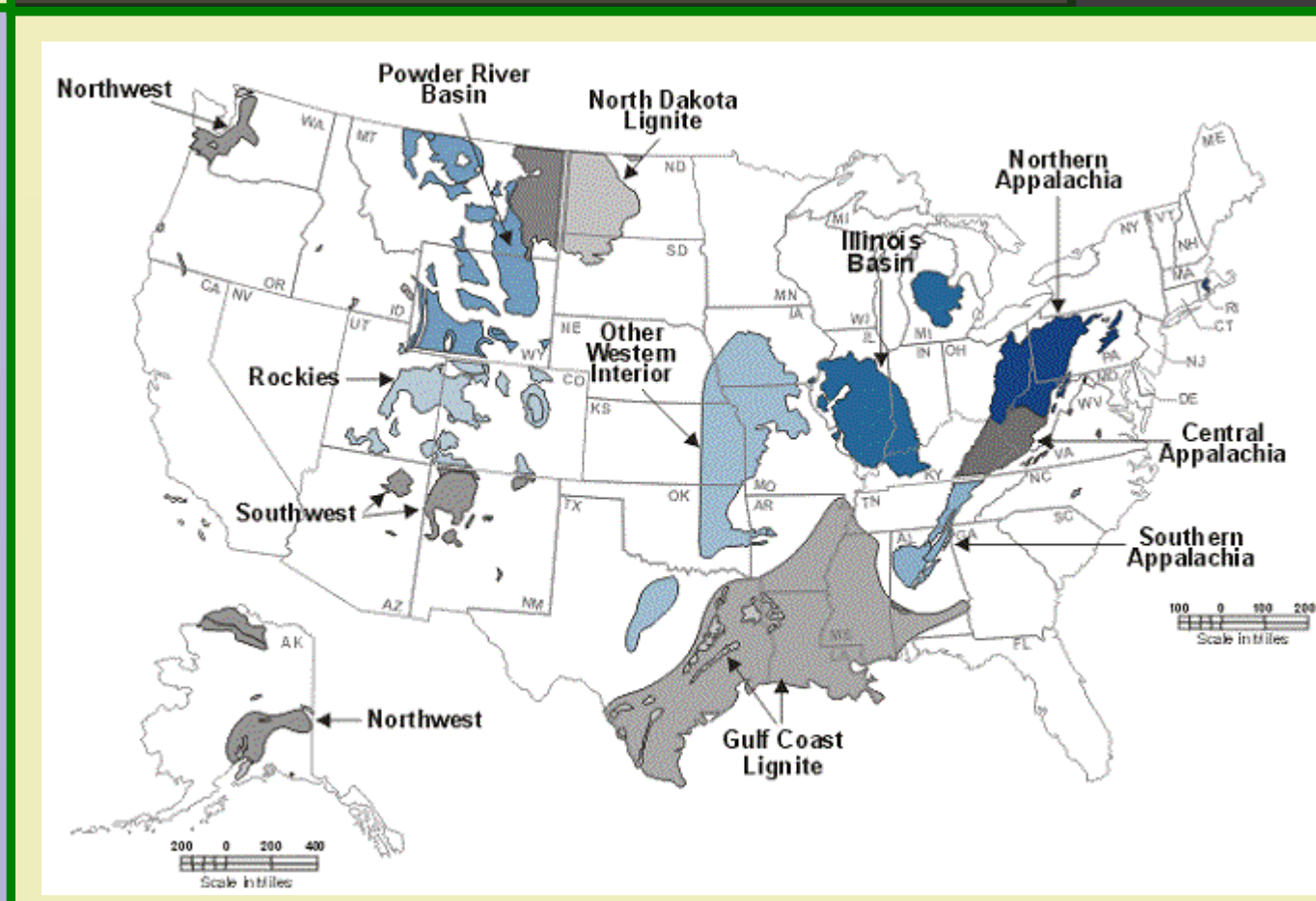
Natural activities that can release mercury into the environment include

- volcano eruptions
- weathering of rocks
- forest fires
- emissions of previously deposited mercury by biologic processes

### Utility Plants Emitting the Most Hg

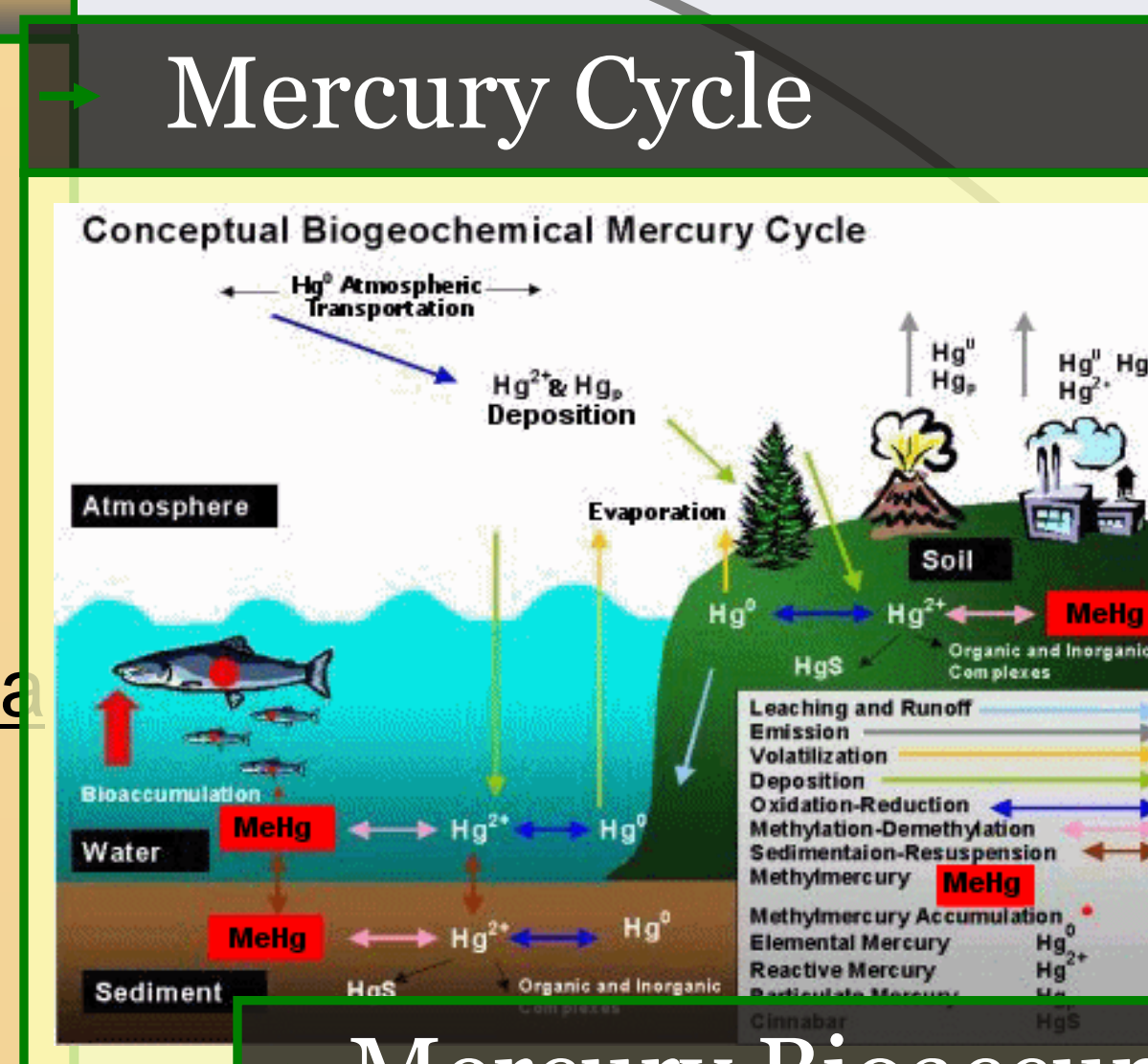


### U.S. Coal Distribution

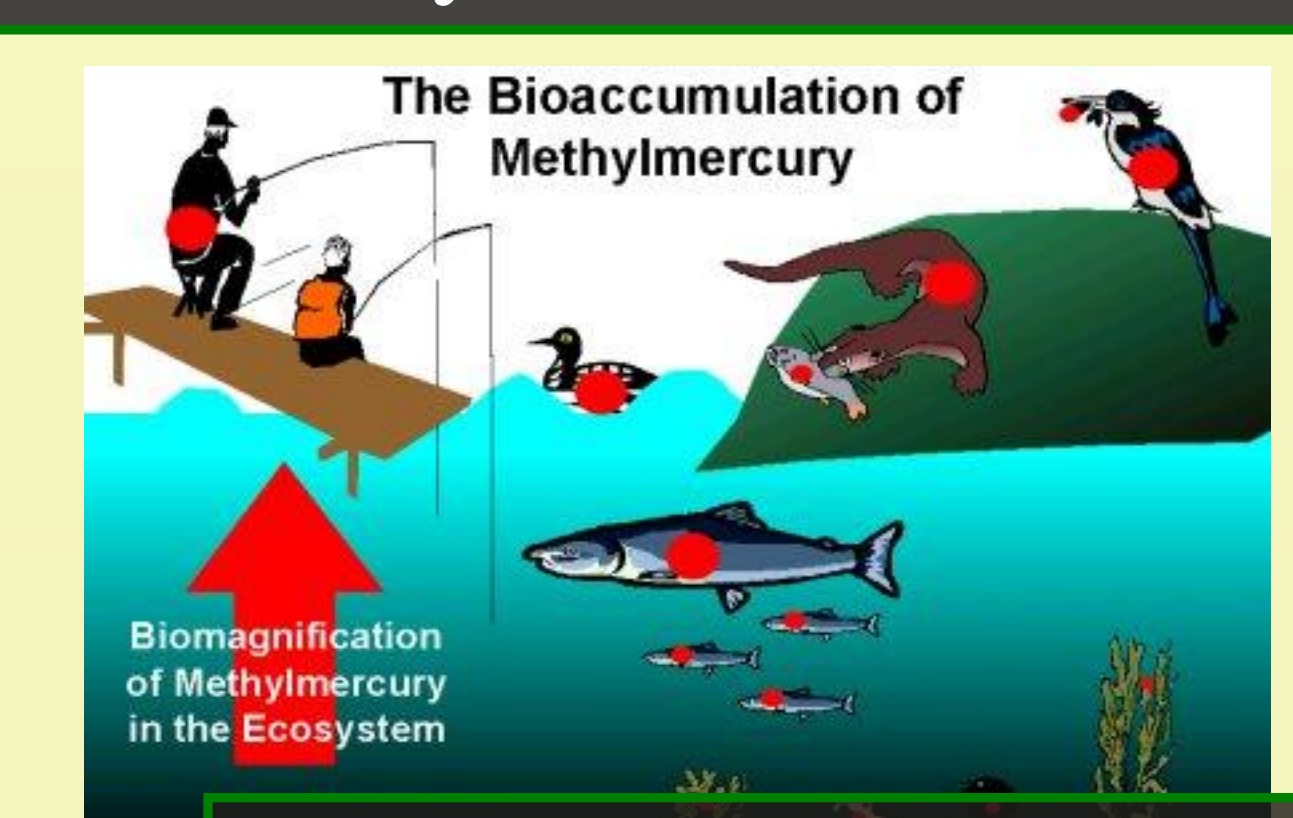


## » Mercury Cycle

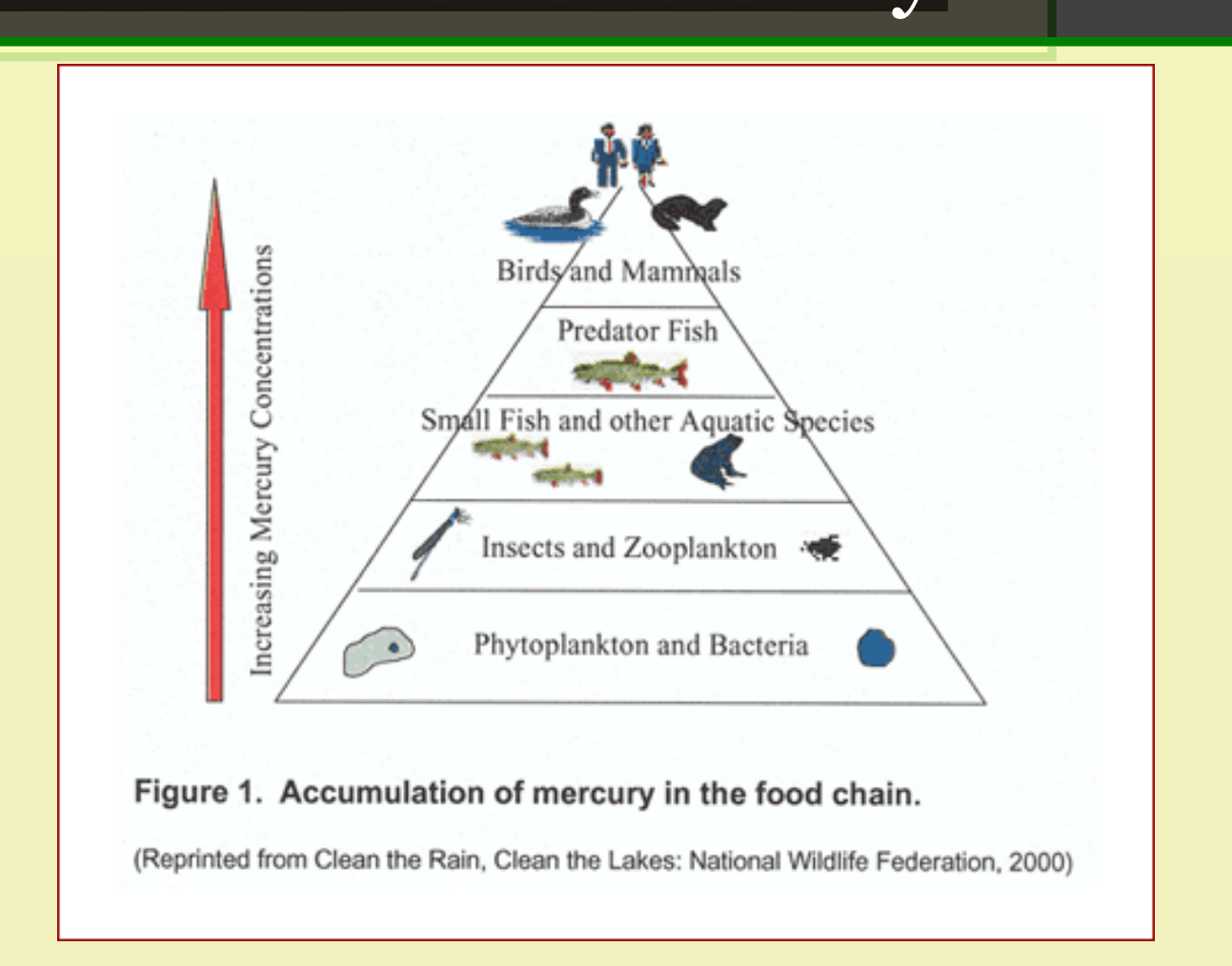
- Mercury is **released into the air** from coal-fired power plant smokestacks and other sources.
- Hg falls to the ground in rain and is **deposited into rivers, lakes and streams** (a process known as deposition).
- As Hg settles into bodies of water, **bacteria convert it to methylmercury**, a highly toxic compound.
- Methylmercury **builds up** (or bioaccumulates) in animals that eat fish (such as pike, bass, tuna, shark and swordfish).
- Methylmercury bioaccumulates in humans.



## » Mercury Bioaccumulation



## » Bioaccumulation Pyramid



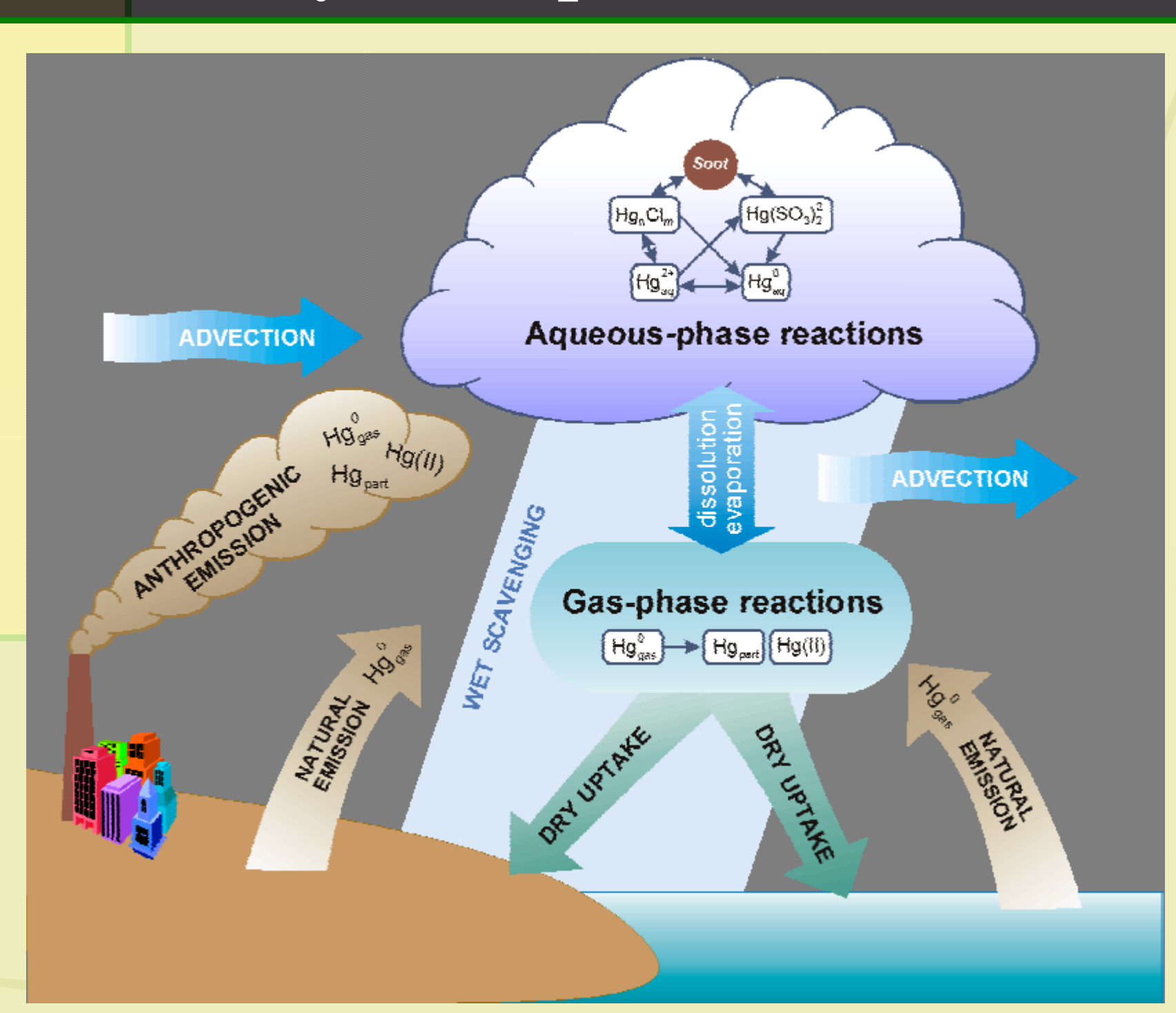
## » Mercury Transport

### Factors

- Form of mercury emitted
- Location of the emission source
- How high above the landscape the mercury is released (e.g., the height of the stack)
- Surrounding terrain
- Weather condition

Atmospheric mercury can be transported over a range of distances before it is deposited, potentially resulting in deposition on local, regional, continental and/or global scales. Mercury that remains in the air for prolonged periods of time and travels across continents is said to be in the "**global cycle**."

## » Mercury Transport Model



## » Mercury Exposure

### Factors

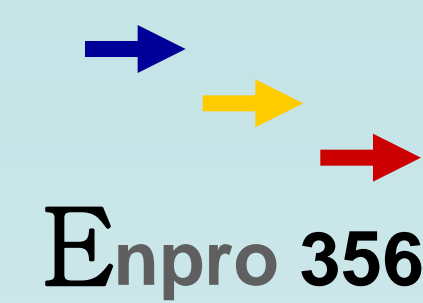
- Dosage - Concentrated consumed
- Duration of exposure - Length of time exposed
- Route of exposure - Means of entering the body
- Age and health of the person
- Chemical form of mercury - elemental (metallic), inorganic compounds, or organic compounds.

## Exposure Methods & Forms

Hg Form	Methyl-mercury	Elemental	Inorganic & Organic
Exposure	consume food with Hg	breathe vapor & absorb through lung; swallow as liquid form	absorb through gastrointestinal tract
Examples	Eating fish or fish-eating animals	Lab spills; Hg product breakage; Dental amalgam	Fungicides; other outdated products
Symptoms	Brain and nervous system; cognitive thinking, memory	tremors; emotional changes; performance deficits	skin rashes; mood swings; memory loss;
Characteristics	organic form of mercury	liquid metal; invisible, odorless toxic vapor	mercury salts; white powders or crystals

## » Health Effects

- Exposure to mercury can be toxic and lethal at high levels.
- Mercury = Developmental **neurotoxin**
- Pregnant women and children are more sensitive
- Mercury can interfere with **brain development** of fetuses and children
- CDC estimates **one in 12 women** of childbearing age in the United States have unsafe levels of mercury in their blood (more than 600,000 newborns each year)
- Safe consumption level:** EPA's reference dose = 0.1 micrograms of mercury per kilogram of body weight per day (**0.1 ug/kg/day**).



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## Regulations

### New Regulations: The Clean Air Mercury Rule (CAMR)

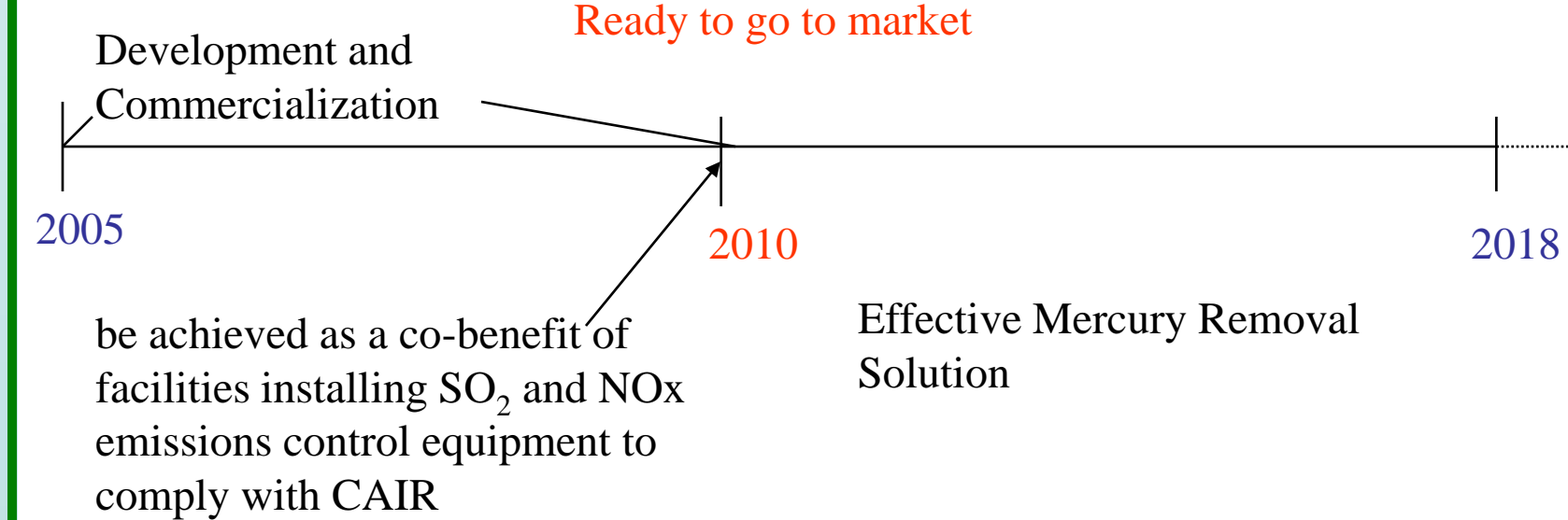
- Cap-and-trade system for reducing emissions
- Limitations of emissions (per generating unit):
  - Bituminous coal:  $21 \times 10^{-6}$  lb/MW·h
  - Subbituminous coal
    - Wet FGD:  $42 \times 10^{-6}$  lb/MW·h
    - Dry FGD:  $78 \times 10^{-6}$  lb/MW·h
  - Lignite coal:  $145 \times 10^{-6}$  lb/MW·h
- 1st phase cap of 38 tons per year in 2010—21% decrease in emissions
- 2nd phase cap of 15 tons per year in 2018—69% decrease in emissions
- CAMR applies to all generators firing more than 73 MW or sell more than 25 MW and more than 1/3 of potential output capacity to any power distribution system
- No provisions for grandfathering

### The Opportunity Ahead

- Any technology that can meet the regulations effectively is needed
- EPA only defined emissions limits and not the technology
- The market has been left open for many technologies to compete and have an opportunity

### 2005-2010-2018

In 2010, 21% reduction of mercury emissions  
In 2018, 70% reduction of mercury emissions.

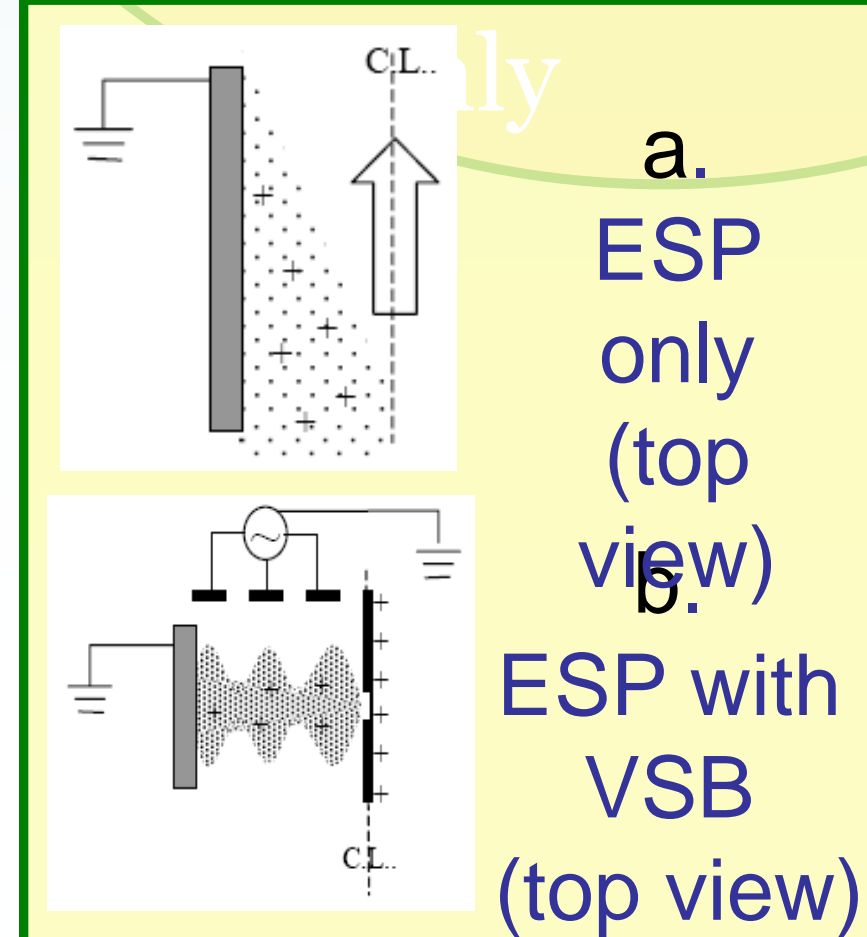


## Competition

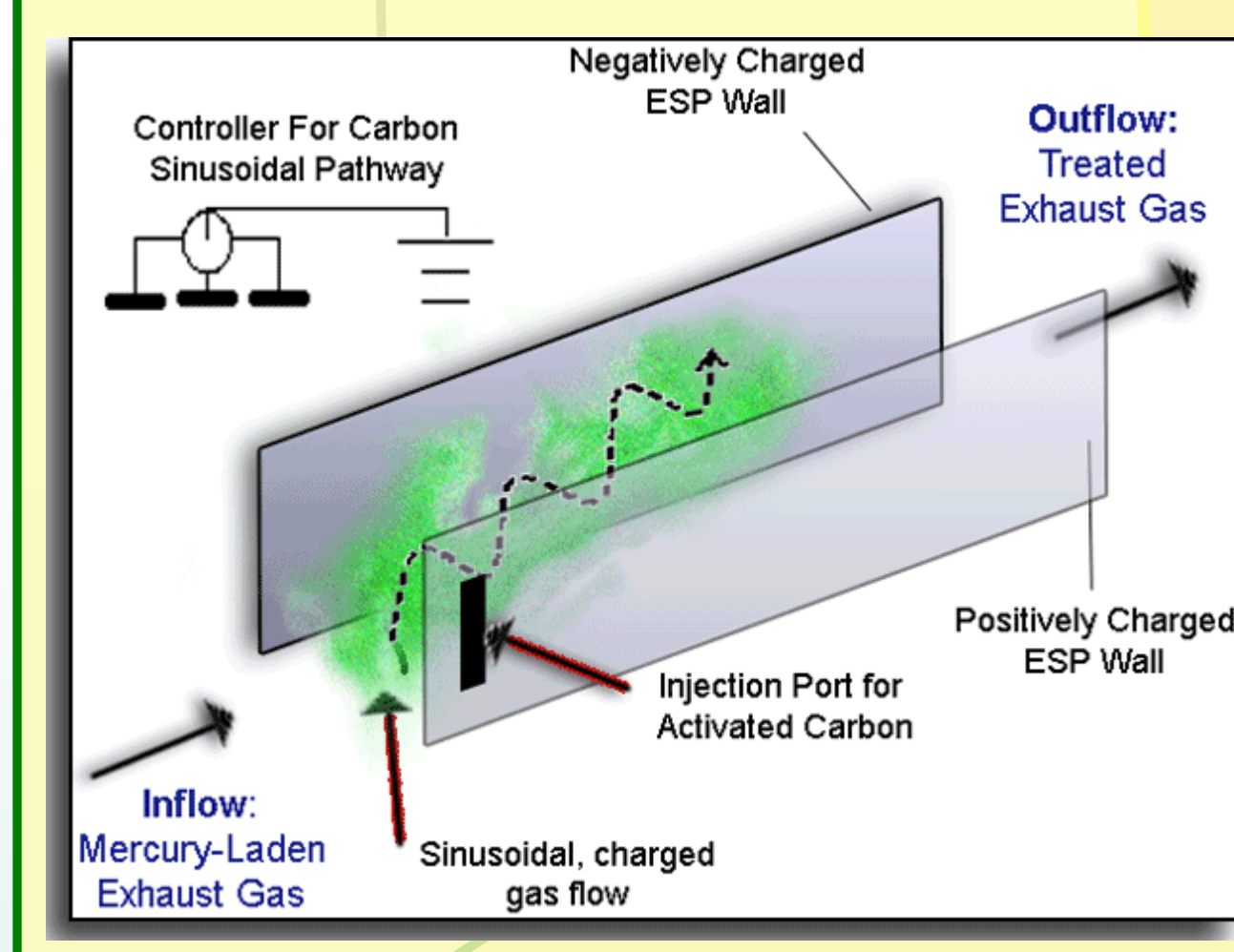
### Mercury Capture Process

- 95% removal of gaseous mercury.
- Amended Silicate™ Technology**  
70%-96% mercury capture
- Advanced Hybrid Filter Technology**  
removes 50-90% of Hg
- Multipollutant Technologies**  
Average of 85% Hg removal efficiency
- Electro-Catalytic Oxidation**  
80 to 90% capture of the mercury

### VSB vs. ESP



### VSB Prototype



## Advantages

### Functional advantages

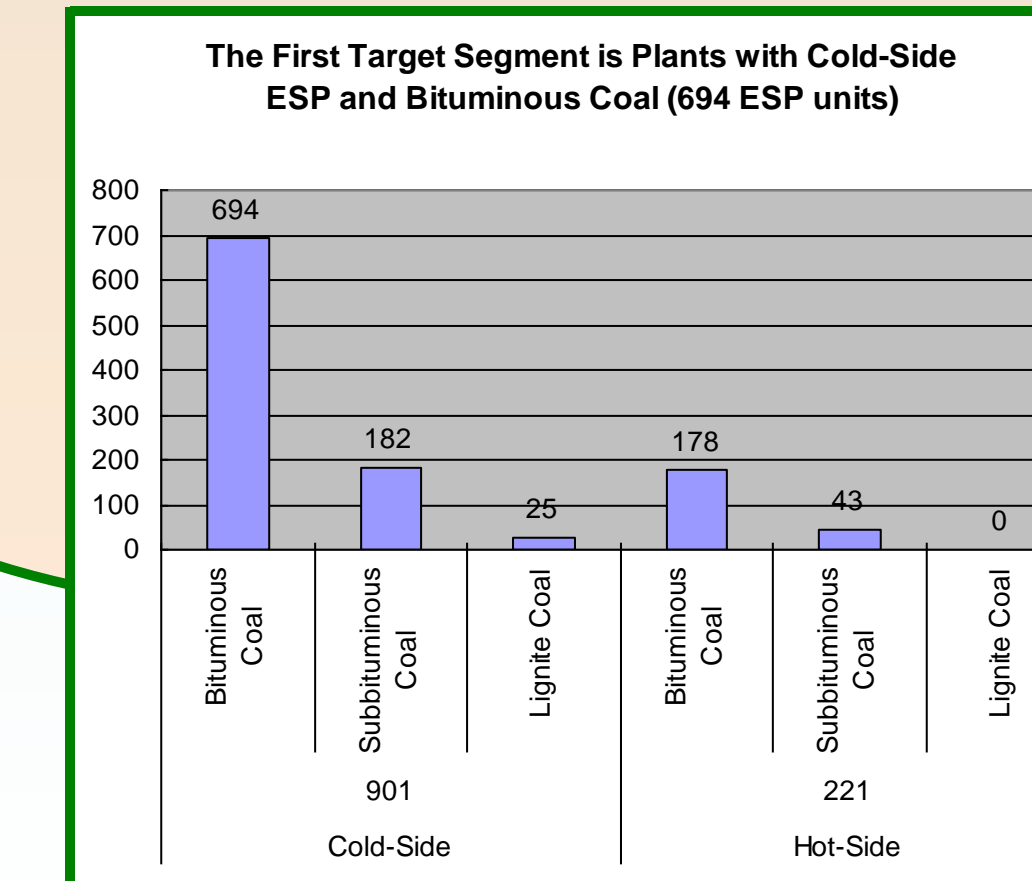
- Dense sorbent loading.
- High gas-particle relative motion.
- No pressure drop.
- Eliminates jet mixing limitations.

### Other advantages

- Sorbent recycling, separate fly ash collection
- Flexible, real-time responsive.
- Safe and transferable

## Market Size

- Clean Air Mercury Rule (CAMR) created a mandatory market for mercury removal technology.
- Coal-fired Power Plants are mandated to pay for solutions.
- The VSB should, once developed, efficiently remove mercury generated by coal-fired generators, and provide an **economical solution** for coal-fired power plants.



## A New Star - VSB

The Virtual Sorbent Beds (VSB) Technology, developed by Dr. Herek Clack, is specifically designed to increase the efficiency of mercury removal from coal combustion exhaust. VSB builds on two proven technologies, Activated Carbon Injection (ACI) and Electrostatic Precipitator (ESP).

### VSB Process

1. Mercury contaminated gas enters the ESP
2. Electrically charged Activated Carbon (AC) is driven in a dense "bed" from one ESP wall to the other.
3. Pathway of AC bed is controlled sinusoidally.
4. Hg contained in the gas is adsorbed onto the AC.
5. The Hg-laden AC is collected
6. Hg-laden AC is either recovered or disposed as a hazardous waste.

### VSB

#### Functional advantages

- Dense sorbent loading.
- High gas-particle relative motion.
- No pressure drop.
- Eliminates jet mixing limitations.

#### Other advantages

- Sorbent recycling, separate fly ash collection
- Flexible, real-time responsive.
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## Decision

### License Charge

- Based on how developed the technology is.
- How proven it is.
- How much more involved the team is after selling/licensing.

### Strategy

- Partner up with a bigger Architectural Engineering Firm
  - Due to low knowledge and resource.
  - Affects how we sell it by changing who we're selling to.

## Financial Data

### Installation Cost

- Needs about a week to install
- Labor = \$2,000 (40 total hours of work)
- No additional steel necessary

### Operation Cost

- 272 MW system was estimated at about \$500,000.
- Power usage (average power plant) 27 million kilowatt a day (14,000 tons of coals to operate)



## Risks

- Technology still under development
- Only have prototype built
- Do not know final size
- Has not been placed in an ESP
- Has not been tested with Hg (mercury)
- Unknown efficiency
- Cost estimates inaccurate
- Have not been developed Hg disposal method
- Entering market late
- Small market
- No market name
- Patent pending

